

1 *Original Article*

2 **Biocompatibility and antibacterial properties of TiCu(Ag) thin films produced by**
3 **physical vapor deposition magnetron sputtering**

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20 **Running title:** Biocompatibility and antibacterial properties of TiCu(Ag) thin films

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26 **Abstract**

27 Mechanical robustness, biocompatibility, and antibacterial performance are key features
28 for materials suitable to be used in tissue engineering applications. In this work, we
29 investigated the link existing between structural and functional properties of TiCu(Ag) thin
30 films deposited by physical vapor deposition magnetron sputtering (MS-PVD) on Si
31 substrates. Thin films were characterized by X-ray diffraction (XRD), nanoindentation,
32 atomic force microscopy (AFM), and X-ray photoelectron spectroscopy (XPS). The TiCu(Ag)
33 films showed complete amorphous structure and improved mechanical properties in
34 comparison with pure Ti films. However, for contents in excess of 20% Ag we observed the
35 appearance of nanometric Ag crystallite. The TiCu(Ag) thin films displayed excellent
36 biocompatibility properties, allowing adhesion and proliferation of the human fibroblasts
37 MRC-5 cell line. Moreover, all the investigated TiCu(Ag) alloy display bactericidal
38 properties, preventing the growth of both *Pseudomonas aeruginosa* and *Staphylococcus*
39 *aureus*. Results obtained from biological tests have been correlated to the surface structure
40 and microstructure of films. The excellent biocompatibility and bactericidal properties of these
41 multifunctional thin films opens to their use in tissue engineering applications.

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44 **Keywords:** bactericidal; biocompatibility; copper; human fibroblast; physical vapor
45 deposition (PVD); silver; titanium; thin films.

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51 **1. Introduction**

52 The aging of world population drives an increasing demand of tissue and organ
53 replacements [1]. More than 10 million transplantations are performed annually, with a yearly
54 increase of about 6% and an overall cost of more than \$500 billion per year [2]. However,
55 tissue and organ transplantations present two major limitations: the low availability of donors
56 and/or risk of disease transmission and immune rejection [3-5].

57 Tissue engineering (TE) is an emerging and promising alternative approach of
58 biomedicine to treat or to replace damaged tissues and organs. TE combines materials science,
59 chemistry, physics, and cell biology to allow tissue and organ repair or reconstruction. TE is
60 often based on nanoscaffolds that enables cell adhesion, migration, proliferation, and
61 differentiation [6-9]; the properties of the scaffold mainly depends upon the types of
62 biomaterial and fabrication techniques [8]. In particular, the size, the shape, and patterning of
63 adhesion sites are crucial elements in the design of effective scaffold surfaces. Moreover,
64 nano-scaffolds must be biocompatible and can be combined with organic and inorganic
65 materials to mimic the structure and function of the natural extracellular matrix (ECM), which
66 allows cells to accomplish the biochemical and biophysical functions related to tissue and/or
67 organs regeneration [8, 10]. Cells grown on nanoscaffolds can generate biocompatible,
68 immunocompatible, and biofunctional tissues inside the body, counteracting the drawbacks
69 associated with autologous grafting and allograft tissue transplantation, thereby alleviating the
70 risk of rejection [8-9].

71 Another critical issue of nanoscaffolds is their capability to prevent microbial growth.
72 Antimicrobial capability is conventionally obtained by means of biochemical approaches
73 relying on nanoscaffold coating with biocidal substances such as silver and antibiotics.
74 Nonetheless, chemical toxicity, antimicrobial durability, and microbial resistance remain
75 critical problems [11].

76 In this complex framework, surface engineering and development of nanostructured
77 thin films is gaining importance, especially for applications where a combination among
78 surface hardness (or wear resistance), biocompatibility, and antibacterial performance are
79 desired. Recently, multi-element thin films, such as Zr-based thin films (*e.g.*, Zr-Cu, Zr-Cu-
80 Ag, ZrCN, Zr/ZrCN multilayer) [12, 13] and Ti-based thin films (*e.g.*, TiN, TiCu, Ti-Zr-Si)
81 have emerged as a new class of nano-engineered thin films, featuring an excellent combination
82 of high mechanical strength and biocompatibility. Moreover, these films are promising
83 systems for biocompatible coating deposition. Indeed, the use of physical vapor deposition
84 (PVD) for their growth allows a fine control of the material nanostructure, which leads to
85 increased hardness and wear resistance [14, 15].

86 The desired combination of mechanical strength, biocompatibility, and antimicrobial
87 activity can also be achieved by constructing multi-layers. Recent studies have confirmed the
88 potential antibacterial behavior of Au, Cu, Zn, Ag additions to Ti-based films [16, 17]. The
89 biocidal performance of Cu is linked to the release of Cu^{+1} and Cu^{+2} ions, as observed in TiCu
90 [18, 19]. Recently, Cu-based systems have also been proposed to engineer surfaces with
91 antiviral properties, also in the framework of the COVID-19 pandemic [20]. Very recent
92 examples include Cu-coated touch surface fabricated by cold-spray technology, as well as
93 antiviral $\text{Cu}_x\text{O}/\text{TiO}_2$ photo catalyst thin films with photo-activated anti-viral properties [21,
94 22]. Consequently, the TiCu systems are of particular interest for the generation of material
95 systems, featuring both anti-infective properties and surface hardness, as required in human
96 implants and/or touch surfaces [23, 24].

97 Here, we report the development of antibacterial metallic TiCu(Ag) PVD sputtered
98 thin films combining biocompatibility with relatively high surface hardness. By using high-
99 resolution surface chemical and morphological characterization, combined with cell growth

100 studies and antibacterial tests, a significant biocompatibility and antibacterial properties have
101 been observed, and correlated to TiCu(Ag) film structure and surface properties.

102

103 **2. Materials and methods**

104 *2.1. Thin film deposition*

105 TiCu(Ag) thin films were deposited on 21×7 mm² coupons extracted from 4" Si (100)
106 undoped wafers by means of direct current (DC) magnetron sputtering in a deposition system
107 equipped with three unbalanced magnetrons. The choice of a silicon substrate eases the study
108 of the process-structure-property correlations for the Ti-Cu-Ag system. Nonetheless, we plan
109 to use other substrates of commoner use in the biomedical realm. We have employed for the
110 deposition Ti, Cu, and Ag targets with a 3" diameter and featuring 99.99% purity. The Si
111 substrates were cleaned in ultrasonic bath and ethanol for 10 minutes before mounting them
112 on the substrate holder. An Ar⁺ sputtering step (powered by radio frequency (RF) power
113 supply at 50 KHz, at the Ar pressure of 1.2 Pa and a discharge power of 0.03 KW) was
114 performed to clean and activate the Si surface immediately prior to metal deposition. The
115 distance between the substrates and targets was 70 mm, while the substrate was kept in rotation
116 at 80 rpm. All the depositions were performed at a 0.52 Pa Ar pressure (chamber base vacuum
117 of 1.0×10⁻⁵ Pa), with no intentional substrate heating. By applying different DC-power to the
118 targets for 40 min, 4 sample sets were obtained, always keeping the Ti:Cu ratio equal to ~1.
119 Deposition conditions, thickness, and composition are listed in Table 1.

120

121 *2.2. Characterization of thin films*

122 Crystallographic structure of the thin films was carried out by X-ray diffraction (XRD),
123 using a θ -2 θ Bruker D8 Advanced system with Cu K α radiation ($\lambda = 0.154$ nm). Diffraction
124 scans were performed by using grazing incident angle of 0.75 deg with time step of 0.02°/sec.

125 The composition of the thin films was estimated via energy dispersive X-ray spectroscopy
126 (EDX, Oxford instrument INCA), using built-in sensitivity factors for calibration. The film
127 thickness was measured by using a white light optical profilometer with a Leica DCM 3D
128 software package via automatic step measurement of the coated and the uncoated parts of the
129 substrate.

130 The elastic modulus (E) and hardness (H) values were determined using nano-
131 indentation testing method using a KLA-Nanomechanics G200 fitted with a Berkovich
132 diamond indenter operating in continuous stiffness measurement mode, hence allowing to
133 obtain both E and H as a continuous function of the depth from a single indentation experiment
134 [25]. A standard fused silica sample was tested before and after a batch of measurements to
135 calibrate the tip, so to ensure the reliability of the results. A least 25 indentations were
136 performed on each sample. Calculations were made by the Oliver and Pharr method from the
137 load-displacement curve using 10% of the film thickness at the maximum indentation depth
138 [26].

139

140 2.3. *X-ray Photoelectron Spectroscopy (SR-XPS)*

141 X-ray Photoelectron Spectroscopy (SR-XPS) measurements were performed at the
142 materials science beamline (MSB) of the Elettra synchrotron radiation source (Trieste, Italy).
143 The UHV end station, with a base pressure of 2×10^{-10} mbar, is equipped with a SPECS
144 PHOIBOS 150 hemispherical electron analyzer and a dual-anode Mg/Al X-ray source, an ion
145 gun, and a sample manipulator with a K-type thermocouple attached to the rear side of the
146 sample. Al $K\alpha$ radiation at 1486.6 eV photon energy impinging at 60° was used to analyze the
147 Cu2p, Ti2p, Ag3d, C1s, N1s and O1s core levels on the respective samples. Photoelectrons
148 were detected at normal emission geometry.

149 Calibration of the energy scale was made referencing the spectra to the C1s core level
150 signal of aliphatic C atoms (285.0 eV). Curve-fitting analysis of the experimental spectra was
151 carried out using Gaussian curves as fitting functions. The Ti2p_{3/2,1/2} core level were fitted
152 using a spin-orbit splitting of 5.7 eV and a (2p_{3/2}/2p_{1/2}) branching ratio of 2; the Cu2p_{3/2,1/2}
153 doublets were fitted using a spin-orbit splitting of 19.8 eV and a (2p_{3/2}/2p_{1/2}) branching ratio
154 of 2. For the Ag3d_{5/2,3/2} doublets, a splitting of 6.0 eV and a branch 3d_{5/2}/3d_{3/2} ratio of 1.5 were
155 used. When different species were identified in a spectrum, the same Full Width at Half
156 Maximum (FWHM) value was set for all individual photoemission peaks. Atomic ratios were
157 calculated from peak intensities by using Scofield's cross section values.

158

159 2.4. *Surface sterilization*

160 Thin films were rinsed in 70% ethanol in sterile deionized water and then flamed with a
161 Bunsen burner. Sterilization was performed under biosafety cabinets with installed HEPA
162 filters to avoid contamination. After sterilization, films were air-dried, and structural and
163 mechanical properties were evaluated in order to verify stability prior to testing for
164 biocompatibility and antibacterial properties.

165

166 2.5. *Human cells culture*

167 MRC-5 human lung fibroblasts were cultured in Dulbecco's Modified Eagle Medium
168 (DMEM) (Corning, VA, USA) complemented with 10% fetal bovine serum (FBS) (BioWest,
169 Nuaille, France), 100 mg/mL penicillin and streptomycin (Merck KGaA, Darmstadt,
170 Germany), and 2.0×10⁻³ M L-glutamine (Merck KGaA). Cells were grown at 37 °C and 5%
171 CO₂. Before seeding, cells were counted using the BLAUBRAND® counting chamber (Brand
172 GMBH, Wertheim, Germany).

173

174 2.6. *Evaluation of surfaces biocompatibility*

175 To correlate MRC-5 cell density to the relative luminescence units (RLU) value, a
176 calibration curve was set up. With this aim, MRC-5 cells were plated in triplicate in opaque
177 96-well plates at a density of 2.5×10^3 , 5.0×10^3 , 1.0×10^4 , 2.0×10^4 , and 4.0×10^4 cells/well. On
178 the basis of the results obtained from the calibration curve, 1.0×10^4 MRC-5 fibroblasts were
179 seeded on sterile TiCu surfaces coated with 0%, 10%, 20%, and 30% Ag thin films placed
180 into opaque 96-well plates. Cell growth and proliferation was assessed by incubating MRC-5
181 fibroblasts for 10 min with the Cell Titer-Glo® Luminescent reagent (Promega, Madison, WI,
182 USA) added in a 1:1 ratio with the complete cell culture medium. Luminescence was measured
183 using the Tecan Spark 10M plate reader (Tecan, Männedorf, Switzerland). Background
184 luminescence was measured in the complete culture medium without cells, and subtracted
185 from each experimental value.

186

187 2.7. *Testing of the antibacterial properties of TiCu(Ag) thin films*

188 Bacteria were routinely grown in Nutrient Broth (NB) No. 2 (# CM0067B; Thermo
189 Scientific™, Waltham, Massachusetts, USA). The day before the experiment, glycerol stocks
190 of *S. aureus* ATCC 25923 or *P. aeruginosa* ATCC 15692 (strain PAO1) were streaked on NB
191 supplemented with 15% agar (NA) plates and incubated at 37 °C for 24 h. By using sterile
192 inoculating loops, bacterial colonies were transferred in 1 mL NB diluted 1:500 (NB_{1:500}) in
193 deionized sterile water and the bacterial concentrations was adjusted to $\sim 5.0 \times 10^7$ colony
194 forming unit (CFU)/mL.

195 The plate count method was conducted according to a minor modification of the
196 International Standard ISO 22196 protocol [27]. Briefly, each sterilized TiCu and TiCu(Ag)
197 surface, as well as the glass control surface, was placed into a Petri dish (\varnothing 3 cm). Then, 0.005
198 mL of a suspension of either *S. aureus* or *P. aeruginosa* at a concentration of $\sim 5.0 \times 10^7$

199 CFU/mL was dripped onto the surfaces and samples were incubated at 37 °C overnight (ON)
200 at 99% relative humidity. After incubation, each surface was placed into 1 mL NB_{1:500} at room
201 temperature for 15 min, and then vortexed for 1 min to allow the detachment of bacteria from
202 the surface. The bacterial suspension was appropriately diluted and plated on NA for CFU
203 counts. To determine the CFU at time 0 h, suspensions of either *S. aureus* or *P. aeruginosa*
204 (presumptive concentration $\sim 5.0 \times 10^7$ CFU/mL) were appropriately diluted in saline and
205 plated onto NA. At least two samples were assessed for each bacterial strain. The antibacterial
206 activity (BA) was calculated by the following formula:

$$207 \quad \text{BA} = [(N_{0\text{h}} - N_{24\text{h}}) / N_0] \times 100\%$$

208 Where, $N_{0\text{h}}$ and $N_{24\text{h}}$ are the CFU average numbers counted at 0 h and 24 h, respectively,
209 for each type of surface.

210

211 2.8. *Ag and Cu ions release from TiCu and TiCu(Ag) thin films in agar*

212 An agar diffusion assay was performed to detect the release of bacterial growth
213 inhibitors. A suspension of either *S. aureus* or *P. aeruginosa* ($\text{OD}_{600} = 0.1$) was uniformly
214 spread onto NA plates using a sterile cotton swab. TiCu and TiCu(Ag) surfaces were placed
215 on the NA inoculated plates. The glass surface was used as negative control, whereas antibiotic
216 discs (*i.e.*, erythromycin, E 15 μg ; and amikacin, AK 30 μg) were used as positive control of
217 bacterial inhibition. After 16 h incubation at 37 °C, the release of antibacterial factors by
218 TiCu(Ag), glass surfaces, and antibiotic discs was visually assessed by the presence of the
219 inhibition zone around the sample.

220

221 2.9. *Ag and Cu ions release from TiCu and TiCu(Ag) thin films in solution*

222 The concentrations of Ag and Cu ions released by TiCu and TiCu-30% Ag thin films
223 were measured by inductively coupled plasma optical emission spectrometry (ICP-OES) using
224 an ICP-OES 710 Varian Spectrometer (Agilent Technologies, Santa Clara, CA, USA). Briefly,
225 TiCu and TiCu-30% Ag thin films were immersed in 5 mL NB_{1:500} supplemented with 0.02%
226 (w/v) sodium azide for 24 h and 7 days at 37 °C. Collected medium were mixed with 5%
227 HNO₃, heated for 1 h at 90 °C, and filtered through a Millipore membrane (pore size 0.45µm)
228 prior to ICP-OES analysis. Concentration of Ag and Cu ions released for each thin film was
229 normalized to the volume of bacterial suspension used in the antibacterial assay.

230

231 2.10. Testing of the antibacterial activity of AgNO₃ and CuSO₄

232 *S. aureus* ATCC 25923 and *P. aeruginosa* ATCC 15692 were incubated for 24 h at a
233 density of ~5.0×10⁷ CFU/mL in a final volume of 2 mL of NB_{1:500} or NB_{1:500} supplemented
234 with 4 µg/mL AgNO₃, or 228 µg/mL CuSO₄, or both salts. After 24 h, 0.05 mL aliquot of the
235 bacterial suspension were diluted in saline and plated on NA for CFU counts. To determine
236 the CFU at time 0 h, suspensions of either *S. aureus* or *P. aeruginosa* (presumptive
237 concentration ~5.0×10⁷ CFU/mL) were appropriately diluted in saline and plated onto NA.
238 The BA of AgNO₃ (µg/mL) or CuSO₄ (µg/mL) was calculated by the following formula.

239

$$BA = [(N_{0\text{ h}} - N_{24\text{ h}}) / N_0] \times 100\%$$

240

241

Where, N_{0 h} and N_{24 h} are the CFU average numbers counted at 0 h and 24 h,
respectively, for each condition assayed.

242

243 2.11. Morphological characterization

244

245

The surfaces morphology was characterized by AFM and optical microscopy; the results
obtained were compared to control samples represented by a suspension of bacteria poured

246 directly on the TiCu surface. AFM measurements were performed using a Dimension ICON
247 AFM (Bruker, Santa Barbara, CA) operating in peak-force mode. The AFM was equipped
248 with a ScanAsyst-Air Bruker silicon probe featuring a nominal cantilever elastic constant of
249 0.4 N m^{-1} and a tip with a nominal radius of 2 nm. For each measurement, height sensor and
250 peak force error images were recorded simultaneously. The AFM images were analyzed and
251 processed with the software Gwyddion [28], applying a first-order flattening. Surface
252 roughness was obtained by measuring the root-mean-square deviation of surface heights on
253 $20 \times 20 \mu\text{m}^2$ images. Optical microscopy images were acquired using a Nikon Eclipse ME600
254 microscope equipped with Nikon DXM1200 digital camera (Nikon, Tokio, Japan).

255

256 **3. Results**

257 *3.1. Structural, morphological, chemical, and mechanical characterization of substrates*

258 The X-ray rocking curves of the as-deposited thin films (see Table 1) are shown in Figure
259 1. The diffraction pattern of the TiCu samples shows only a broad band in the $[38-45^\circ]$ 2θ -
260 range, pointing to an amorphous structure [29-30].

261 For Ag concentration exceeding 10%, we observed the appearance of a second feature,
262 whose angular position is compatible with the Ag(111) reflection, showing an increasing
263 intensity and a narrowing for increasing Ag content. This point out to a Ag precipitation
264 phenomenon with the formation of Ag nanocrystallites at Ag contents $> 20\%$. By using the
265 Scherrer's equation on XRD spectra, an Ag crystallite size of the order of 3-5 nm for the film
266 with 20% Ag addition was estimated. No impact on the XRD spectra has been observed
267 flaming with alcohol.

268 The AFM analysis revealed crack-free smooth surfaces for all thin films with an increase
269 in the average roughness as the Ag content increased from 10 to 30% as shown in Figure 2.
270 The amorphous TiCu and TiCu(Ag) thin films with the Ag content below 20% exhibited an

271 average surface roughness well below 0.5 nm. For TiCu(Ag) film featuring Ag content > 20%,
272 we observed the appearance of nanoclusters, that, in agreement with the XRD evidence above
273 discussed, was attributed to the diffusion and segregation of Ag and Cu atoms, leading to the
274 formation of Ag nano-crystalline precipitates. Consequently, the surface roughness increases
275 up to ~1 nm in the Ag-richest sample (see Fig.3). It is worth noting that the surface clusters
276 observed by AFM have a size larger than that obtained by the XRD analysis. This is not in
277 contrast with our hypothesis, since the surface cluster might be formed by the assembly of
278 different nanocrystallites of a lesser size.

279 Thin films with such a low average surface roughness are in general very favorable for
280 antibacterial and biomedical applications [31], especially if antimicrobial agents such as Cu
281 and Ag are added into the protective thin film, which induces a release of metallic ions after
282 exposure to a humid environment, as we shall further discuss in the following sect. 3.4 [32].
283 The elastic modulus E and hardness H of thin films were calculated as a function of Ag
284 contents by the nanoindentation method. The TiCu film exhibited the highest E = 124.3 GPa
285 and H = 7.83 GPa values. As we can observe in Figure 3A and 3B, the addition of Ag into the
286 TiCu(Ag) thin films induced a decrease of both modulus and hardness, with the lowest values
287 of modulus (109.25 GPa) and hardness (6.45 GPa) observed in the sample containing 30% of
288 Ag. The very good adhesion of the coating layer has been addressed in a previous article [33].

289 After the flame sterilization test, no change was observed in the mechanical properties
290 of thin film, consistent with a previous report [33]. All the samples were analyzed by XPS
291 spectroscopy before and after flaming with alcohol (flamed samples will be labelled (F) in the
292 following text). The measured binding energies (BE, ± 0.2 eV), FWHM, and atomic ratios
293 calculated from peak areas for all the analyzed samples are reported in Table S1.

294 In Figure 4 is reported as an example the Ti2p, Cu2p, and Ag3d spectra and the relative
295 curve-fitting analysis for the sample TiCu-30%Ag (F). The measured BE value of the Ti2p_{3/2}

296 signal (458.7 eV) corresponds to the expected value for TiO₂ [34]: indeed, when exposed to
297 air, Ti is always oxidized to titania in the outmost surface layer [35]. The Cu2p_{3/2} signal results
298 from two components peaks located at 933.0 eV and 935.0 eV that have been attributed to
299 metallic (Cu) and oxidized (CuO) copper, respectively. Moreover, the presence of a shake-up
300 satellite, evident in the spectrum at about 943.5 eV, is a distinctive feature of Cu in the +2
301 oxidation state [36]. Finally, the Ag3d_{5/2} main peak position at 368.5 eV is typical of metallic
302 Ag; a small higher BE component, about 10% of the main component peak, located at about
303 369.9 eV can also be observed in the spectrum, and can be attributed to oxidized, positively
304 charged silver atoms, indicated as Ag₂O in Table S1 [37]. In all samples, Ti was completely
305 oxidized to Ti(IV), Cu was partially oxidized to Cu(II), and most Ag was predominantly in
306 the unoxidized metallic state, consistent with the different reactivity towards oxygen of the
307 three metals (Table S1). Since XPS spectroscopy probes a 1-5 nm thin outer layer of the
308 surface, the appearance of O-related signal can be attributed to surface contamination during
309 the sample transport from the growth reactor to the XPS analysis chamber at ELETTRA (not
310 connected *in-vacuo*).

311 Peak areas have been used to calculate atomic ratios between the oxidized and metallic
312 components and between the three elements present on the sample surface (Table S1). The
313 measured atomic ratios between copper and titanium disagree with the expected value of 1:1,
314 possibly because XPS is a surface-sensitive technique with a sampling depth of approximately
315 ~5 nm. Therefore, we argue that the discrepancy observed between measured and expected
316 elemental composition is relative only to the outmost sample surface. The effect could be due
317 to the induced segregation between the two metals, given by their difference in density.

318 For TiCu samples, the Cu content of the sample surface was lower than expected. On
319 the other hand, the Ag:Ti atomic ratio on the surfaces of TiCu(Ag) thin films was higher than
320 those of the targeted bulk values, with a slight saturation effect has been observed at 30% Ag

321 concentration. We suggest that, when Ag was introduced in the mixture, Cu and Ag form an
322 alloy and migrate together to the outmost sample surface, with Ag forming clusters as
323 evidenced by AFM and XRD analysis (see Figures. 1 and 2). We notice that the flaming
324 procedure does not affect the sample surface composition and the oxidation state of the three
325 elements.

326

327 *3.2. Biocompatibility studies of the TiCu(Ag) surfaces*

328 The TiCu(Ag) thin films biocompatibility has been evaluated by measuring human
329 MRC-5 fibroblasts viability based on adenosine triphosphate (ATP) production under aerobic
330 conditions, which reflects the presence of metabolically active cells (Figure 5A). With this
331 purpose, we first determined the correlation between the number of viable cells and the emitted
332 luminescence (relative luminescence unit, RLU) (Figure 5B). Basing on obtained results, we
333 next seeded 1.0×10^4 MRC-5 cells on polystyrene cell culture plates (control) and on sterile
334 TiCu surfaces coated with 0%, 10%, 20%, and 30% Ag (Figure 5C). As shown, the RLU value
335 after 24 h from seeding was comparable among control, TiCu, and TiCu(Ag) thin films, thus
336 indicating that the tested surfaces did not affect cell viability and proliferation. Further, an
337 unaltered morphology of MRC5 cells grown on TiCu and TiCu(Ag) thin films was observed,
338 thus supporting the excellent biocompatibility of the developed thin films (data not shown).

339 Next, we evaluated if the trypsin-mediated detachment of cells affected the
340 biocompatibility of TiCu and TiCu(Ag) thin films. Results obtained indicated that
341 trypsinization did not alter the capability of the TiCu(Ag) thin films to allow MRC-5 growth,
342 supporting their possible use for tissue culture studies (Figure 5D).

343

344 *3.3. XPS analysis of TiCu(Ag) surfaces after human MRC-5 cells trypsinization*

345 To evaluate the possibility of re-use the thin films for biocompatibility test after human
346 cells detachment by trypsinization, TiCu(Ag) surfaces were analyzed by XPS and C1s, N1s,
347 O1s, Ag3d, Ti2p and Cu2p core levels were investigated. XPS spectra and data (BE, FWHM,
348 and atomic ratios) are reported in Figure 6 and Table S2. As already evidenced for the pristine
349 samples, the Ti2p spectra are typical of TiO₂. The Cu2p spectra revealed the presence of both
350 metallic and oxidized (CuO) copper, while the Ag3d signal is typical of metallic silver with a
351 very small component related to oxidized silver (spectra not shown). No relevant change were
352 evidenced on the oxidation state of metals on the sample surface. C1s, O1s and N1s spectra
353 revealed the presence of organic molecules, particularly peptides, deposited on the TiCu(Ag)
354 film surfaces.

355 The curve fitting analysis of C1s core level spectra (Figure 6A) showed four
356 components: (i) the peak at 285.0 eV indicating aliphatic C–C carbons; (ii) the peak at ~286.5
357 eV, related to C–N and C–O carbons of peptide backbones; (iii) the peak at 288.3 eV due to
358 O=C–N peptide carbons; and (iv) the peak at 290.0 eV due to COOH carbons. The N1s
359 spectrum (Figure 6B) comprised a peak at 399.0 eV due to C=N nitrogens, a main peak at
360 400.4 eV related to peptide nitrogens, and a peak at about 402.5 eV due to protonated nitrogens
361 [32, 38]. The O1s spectra (Figure 6C) showed four component peaks: (i) the peak at 530.1 eV
362 assigned to the oxygens of titania; (ii) the peak at about 532.0 eV, assigned to O=C oxygens
363 of the peptide backbone; (iii) the peak at 533.5 eV assigned to C-O oxygens; and (iv) the peak
364 at nearly 535.0 eV, related to physisorbed water [32].

365 In summary, the XPS data analysis points to the presence of peptide residues on the
366 samples after human cells detachment, possibly representing cells residues of previous MRC-
367 5 cell growth not completely removed by trypsin from the thin films, and/or trypsin residues
368 adsorbed on the sample surface. However, the presence of these peptide residues did not
369 influence MRC-5 adhesion and proliferation, as reported in Figure 5D.

370

371 3.4. Evaluation of the bactericidal properties of the TiCu(Ag) surfaces

372 The bactericidal property of TiCu and TiCu(Ag) thin films was evaluated by testing the
373 growth of two well-known nosocomial pathogens, *i.e.* *S. aureus* and *P. aeruginosa*. The
374 experimental protocol is illustrated in Figure 7A. Briefly, *S. aureus* and *P. aeruginosa* were
375 dispersed on either a glass surface or TiCu thin films coated with 0%, 10%, 20%, and 30%
376 Ag. Bacteria were incubated for 24 h at 37 °C in a controlled 99% humidity chamber prior to
377 mechanical detachment and CFU counting. For both species, limited or no loss of bacterial
378 viability was observed on glass after 24 h (Figure 7B). Conversely, no colony growth was
379 observed when bacteria were dispersed on TiCu and TiCu(Ag) surfaces, indicating a strong
380 bactericidal activity (BA) (Figure 7B).

381 To evaluate the diffusion of antibacterial agent(s) in solid media, a suspension of *S.*
382 *aureus* or *P. aeruginosa* was spread over NA plates, and then TiCu(Ag) surfaces were placed
383 onto the inoculated plates. The glass surface was used as the negative control (no growth
384 inhibition), whereas antibiotic discs (*i.e.*, erythromycin, E 15 µg, and amikacin, AK/30 µg)
385 were used as the positive control (growth inhibition due to antibiotic diffusion around the
386 discs). No growth inhibition was observed around both glass and TiCu(Ag) surfaces,
387 indicating no diffusion of inhibitory agent(s) as opposed to the large inhibition halo around
388 the antibiotic discs (Figure 7C).

389 To exclude incomplete detachment of *S. aureus* and *P. aeruginosa* from the NB_{1:500}
390 washed TiCu surfaces, both optical microscopy (Figure 8A) and AFM analyses (Figure 8B)
391 were performed. Microscopy results were compared with positive control samples in which
392 the same number of bacteria was directly dispensed onto the TiCu surfaces, without
393 subsequent washing step. A complete detachment of bacterial cells from TiCu surfaces was
394 observed for both strains. Indeed, the large-scale optical microscopy images showed that on

395 control samples many “coffee-stain” clusters were present, while no clusters were observed
396 on washed TiCu surfaces (Figure 8A). This was confirmed by acquiring several images across
397 the surfaces. By AFM characterization, the clusters on control samples were univocally
398 identified as bacterial aggregates (Figure 8B). Notably, *S. aureus* bacteria showed a round
399 shape, with an average diameter and height of 950 nm and 530 nm, respectively. The *P.*
400 *aeruginosa* bacteria were instead shallower with an average height of 230 nm and showed an
401 elongated shape, often with flagella, having an average long-axis length of 1300 nm. The
402 tendency of bacteria towards clustering in control samples (Figure 7B) was observed for
403 similar deposition conditions and attributed to the capillary flow induced by the evaporation
404 of the drop deposited [39, 40]. Conversely, the TiCu surfaces analyzed after bacteria detaching
405 showed morphological features with heights below 30 nm (thus more than one order of
406 magnitude lower than typical bacteria heights), likely to be attributed to residual of the nutrient
407 medium used for the growth. The root mean square (Rms) surface roughness, as evaluated on
408 $20 \times 20 \mu\text{m}^2$ images, was 1.71 nm and 1.87 nm after detachment of *S. aureus* and *P. aeruginosa*,
409 respectively, while it exceeded 50 nm for the control surfaces with bacteria adhered. This
410 comparative morphological analysis confirms that: (i) bacteria were completely detached with
411 the experimental protocol used for the plate count experiments and, (ii) TiCu(Ag) surfaces
412 were bactericidal for both *S. aureus* and *P. aeruginosa*.

413 To demonstrate that the antibacterial activity of TiCu and TiCu(Ag) thin films was due
414 to the release of Ag and Cu ions into the medium, thin films were immersed for either 24 h or
415 7 days in NB_{1:500} and the concentration of each metal was determined by ICP-OES. Results
416 obtained indicated that after 24 h the presumptive concentration of Cu ions released in the
417 medium from TiCu and TiCu-30% Ag thin films was $228.80 \pm 18.67 \mu\text{g/mL}$ and 227.93 ± 16.59
418 $\mu\text{g/mL}$, respectively (Figure 9A). The presumptive concentration of Ag ions released in the
419 medium after 24 h incubation by the TiCu-30% Ag was $4.07 \pm 0.47 \mu\text{g/mL}$ (Figure 9A).

420 Interestingly, the concentrations of both metals in the medium did not increase after 7 days
421 incubation (Figure 9A). To correlate the antibacterial properties of TiCu(Ag) thin films with
422 the presumptive concentration of released metals, 5.0×10^7 CFU/mL of either *S. aureus* or *P.*
423 *aeruginosa* were incubated for 24 h in NB_{1:500} and in NB_{1:500} supplemented with 4 µg/mL
424 AgNO₃, or 228 µg/mL CuSO₄ or both salts. The addition of AgNO₃, or CuSO₄, or both salts
425 to the medium caused the 100% killing of both *S. aureus* and *P. aeruginosa* (Figure 9B). These
426 results demonstrated that the antimicrobial activity of TiCu and TiCu(Ag) thin films is due the
427 release of Ag and Cu ions into the medium, in agreement with previous suggestions [32].

428

429 **4. Discussion**

430 In this article, the biocompatibility and the antibacterial properties of TiCu and TiCu(Ag)
431 thin films produced by PVD magnetron sputtering method are reported. All the TiCu(Ag) thin
432 films allow adhesion and growth of human MRC5 fibroblast, together with a strong
433 antibacterial activity against both *S. aureus* and *P. aeruginosa*, here considered as prototypes
434 of nosocomial bacterial pathogens. Results are extremely significant for TE applications as
435 biocompatibility and antibacterial performance represent key design parameters for
436 biomaterials. In fact, tissue and implant-associated bacterial infection are a growing problem
437 responsible for increased morbidity and mortality, together with enormous economic losses to
438 the public health system. Bacteria can tightly adhere to the biomaterial surface, and the
439 formation of biofilm can help bacteria to escape the host immune system and antibiotics. In
440 turn, this causes the emergence of bacterial resistance to antibacterial drugs and finally
441 determines implantation failures [41]. Therefore, implanting materials that combine the
442 capability to favor eukaryotic cells adhesion and proliferation for tissue regeneration with
443 strong antibacterial properties are urgently needed. Metallic elements (*e.g.*, Au, Ag, Cu, and
444 Zn) have been proven to exert antibacterial activity by surfaces coating or doping [42].

445 The XPS analysis shows that TiO₂ represents the main oxidation status for titanium,
446 which allows an excellent growth of eucaryotic cells and explains the observed
447 biocompatibility. At the same time, all films show the presence of a combination between
448 metallic and oxidized copper, with an increase of Cu/Ti ratio after addition of Ag. In addition,
449 Ag is shown to be predominantly present in the metallic state with low levels of Ag₂O on the
450 surface. These observations allow to explain the relevant antibacterial performance of the
451 films. In fact, the presence of a single-phase film with homogeneous surface distributions of
452 Cu and/or Ag can be very effective in protecting the surface against bacteria, while
453 maintaining a high biocompatibility thanks to the concurrent homogenous distribution of TiO₂.

454 The biocompatibility of TiCu(Ag) thin films here reported agrees with the fact that Ag
455 displays a very low cytotoxicity when tested in human blood, adenocarcinomic alveolar basal
456 epithelial cells, liver cancer cells, breast cancer cells, and gastric cancer cells [43-45]. On the
457 other side, Cu is a transition metal and is an essential micronutrient in humans. Indeed, Cu is
458 involved in many biosynthetic and metabolic processes, being a cofactor of many redox
459 enzymes and playing a role in iron metabolism, as well as in immunity [46]. It has been
460 reported that Cu exerts a very low cytotoxicity in human cells (TC₅₀ value of 344±4.4 μM in
461 human gingival fibroblast [47]). Here, we tested the TiCu(Ag) thin films biocompatibility
462 using MRC-5 human fibroblasts, a cell line widely used to perform cytotoxicity assays [48-
463 56]. Human fibroblasts are responsible for synthesizing and depositing ECM components,
464 mediate epithelial-mesenchymal interactions allowing other cells to settle and migrate along
465 a three-dimensional support and thereby generating an organ-specific architecture.
466 Noteworthy, fibroblast cells contain high levels of fibronectin and fibrin, which generates
467 rapid and secure fibroblast adhesion to the Ti substrate [53]. Therefore, studying fibroblast
468 adhesion and proliferation on nanosurfaces may be used with the aim of increasing tissue
469 repair in several conditions like healing of acute and chronic wounds [49].

470 Additionally, we demonstrated that the anti-bacterial activity of TiCu and TiCu(Ag) thin
471 films is due to the release of Ag and Cu ions into the medium. Noteworthy, several data
472 support the anti-bacterial, anti-viral, anti-biofilm, and anti-inflammation activity of Ag,
473 especially at nanoscale [43-45, 57, 58]. The antibacterial property of Ag relies on its ability to
474 form pores and penetrate the bacterial wall by reacting with the peptidoglycan component.
475 Indeed, owing to electrostatic attraction and affinity to sulfur proteins, silver ions can adhere
476 to the cell wall and the cytoplasmic membrane. In turn, adhered ions can enhance the
477 permeability of the cytoplasmic membrane. Once into the bacterial cells, silver ions can inhibit
478 cellular respiration, resulting in the generation of reactive oxygen species (ROS), and
479 interrupting metabolic pathways by inhibition of ATP production. ROS can also induce cell
480 membrane disruption and DNA oxidation. Since sulfur and phosphorus are important
481 components of DNA, the interaction of silver ions with DNA can impair its replication, hence
482 cell duplication, ultimately causing bacterial death. Finally, Ag ions can inhibit the synthesis
483 of proteins by denaturing ribosomes in the cytoplasm [59-61]. While low concentrations of
484 Cu are essential for bacteria metabolism, high concentrations, cause cell growth inhibition or
485 even cells death [62, 63]. Therefore, Cu represent an optimal metal to prepare antibacterial
486 titanium alloys [42, 64, 65].

487

488 **5. Conclusion**

489 In this paper, the surface structural and functional properties of amorphous TiCu(Ag)
490 thin film have been investigated. The increase of Ag content is accompanied by the appearance
491 of Ag-nanocrystallites and by a decrease of both elastic modulus and hardness of the thin
492 films. TiCu(Ag) thin films allowed a very good adhesion and growth of fibroblast MRC5 cells
493 irrespective to different Ag content.

494 Based on the multi-technique characterization and cellular studies, it can be concluded
495 that binary TiCu and TiCu-10% Ag showed the best mechanical properties with amorphous
496 glassy structure combined to excellent biocompatibility and antibacterial activity. On the
497 contrary, ternary TiCu(Ag) thin films with 20% Ag content showed moderate mechanical
498 properties, although they display excellent biocompatibility and antibacterial properties.

499 Although many coatings and modified surfaces provide similar antibacterial activity,
500 such surfaces functionalisation can have a detrimental effect on tissue biocompatibility,
501 impairing the integration of the implants into the surrounding tissue. The excellent
502 biocompatibility and bactericidal properties of our multifunctional thin films opens to their
503 use in TE applications. Moreover, these types of thin films could be also used to coat surgical
504 tools and hospital furnishing. For the future, it might be of interest to study these thin films
505 for the biocompatibility of mesenchymal stem cells, one of the most studied stem cells in the
506 TE field due to their great potential to enhance tissue regeneration thanks to the capability to
507 differentiate into cartilage, bone, fat, muscle, tendon, skin as well as hematopoietic-supporting
508 stroma and neural tissue [66]. For the future, it could be also tested if these kinds of thin films
509 could be suitable for applications against SARS-CoV-2 infection [20], since Cu and its oxide
510 have been demonstrated to act as efficient antiviral agents [20, 21, 67].

511

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524

525

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- 709
- 710

711 **Figure captions**

712 **Figure 1. X-ray diffraction pattern.** Diffraction scans of the TiCu and TiCu(Ag) thin films.

713

714 **Figure 2. Characterization of the surface morphology.** Atomic force microscopy (AFM)

715 images of the TiCu and TiCu(Ag) thin films with an Ag content of 10%, 20%, and 30%.

716

717 **Figure 3. Comparison of elastic modulus and surface average roughness (A) and of**

718 **hardness and surface average roughness (B) as a function of Ag content in TiCu thin**

719 **films.**

720

721 **Figure 4. XPS spectra.** (A) Ti2p, (B) Cu2p, and (C) Ag3d core level spectra of sample TiCu-

722 30%Ag and related peak fitting analysis.

723

724 **Figure 5. Evaluation of the biocompatibility of the TiCu and TiCu(Ag) surfaces.** (A)

725 Overview on the cell survival assay used to evaluate MRC5 cells grown on the TiCu surfaces.

726 The Cell Titer-Glo[®] reagent added to the cells generates a luminescent signal (relative light

727 unit, RLU) that is proportional to the ATP present, which in turn is directly proportional to the

728 number of metabolically active cells grown on the TiCu surface. (B) Set up of the assay

729 sensitivity using MRC5 cells grown on TiCu surfaces. Cells were seeded in triplicate at a

730 density of 2.5×10^3 , 5.0×10^3 , 1.0×10^4 , 2.0×10^4 , and 4.0×10^4 cells/well. (C) Cell viability assay

731 measured 24 h after from seeding 1.0×10^4 MRC5 cells in triplicate on the 96-well plastic

732 surface (Ctrl) or on the TiCu surfaces coated with 0%, 10%, 20%, and 30% Ag. (D) Cell

733 viability assay measured after 24 h from seeding 1.0×10^4 MRC5 cells in triplicate on the TiCu

734 surfaces coated with 0%, 10%, 20%, and 30% Ag, which were either previously treated with

735 trypsin or not. Background luminescence was measured in well containing the medium

736 without cells and subtracted from experimental values. Data are reported as means \pm SD
737 (Student's t test, $p < 0.05$).

738

739 **Figure 6. XPS spectra after human cells detachment by trypsinization.** C1s (A), N1s (B)
740 and O1s (C) XPS spectra of TiCu(Ag) films after human MRC-5 fibroblasts detachment with
741 trypsin and related peak fitting analysis.

742

743 **Figure 7. Bactericidal properties of the TiCu and TiCu(Ag) surfaces.** (A) Schematic
744 representation of the protocol used to test the bactericidal activity of the TiCu(Ag) surfaces.
745 (B) Bactericidal effect of TiCu(Ag) surfaces determined after 24 h incubation at 37°C with
746 99% humidity of *S. aureus* and *P. aeruginosa* cells on TiCu(Ag) surfaces. The glass surface
747 was used as negative control for bacterial killing. After incubation, aliquots of the bacterial
748 suspension were diluted in saline and plated on agar for colony forming unit (CFU) counts.
749 (C) Plate test for release of antibacterial agents from TiCu(Ag) surfaces. *S. aureus* and *P.*
750 *aeruginosa* were used as test species. Antibiotic discs and glass were used as positive and
751 negative controls of bacterial inhibition.

752

753 **Figure 8. Surface morphology of TiCu samples after bacteria detachment.** (A) Optical
754 microscopy images and (B) AFM topographic (top panels) and peak-force error (bottom
755 panels) images of control (Ctrl) and TiCu surfaces on which *S. aureus* or *P. aeruginosa* were
756 grown. Images were acquired after bacteria detaching and were compared to control samples
757 where a suspension of each bacteria strain, containing the same number of bacteria as in the
758 experiment, was poured directly on the TiCu surface. In the insets shown in panels B, 3D
759 zoom ups are displayed.

760

761 **Figure 9. Ag and Cu ions release from TiCu and TiCu-30% Ag thin films and**
762 **bactericidal activity.** (A) The concentration of Ag and Cu ions released from TiCu and TiCu-
763 30% Ag thin films, immersed for either 24 h or 7 days in the standard test medium (NB_{1:500}),
764 was determined by inductively coupled plasma optical emission spectrometry (ICP-OES). The
765 reported final concentrations of Ag and Cu ions released for each thin film were normalized
766 to the volume of bacteria used in the antibacterial assay. The standard test medium (NB_{1:500})
767 was used as control. ND, not detectable. (B) Bactericidal effects of the presumptive
768 concentration of Ag and Cu ions released from the thin films. *S. aureus* and *P. aeruginosa* cells
769 were incubated in NB_{1:500} or NB_{1:500} supplemented with 4 µg/mL AgNO₃, or 228 µg/mL
770 CuSO₄, or both salts. After 24 h incubation, aliquots of the bacterial suspension were diluted
771 in saline and plated on agar for colony forming unit (CFU) counts.

Table 1. Composition of the TiCu(Ag) thin films.

Film composition	Ti:Cu ~ 1	Power at Ti (W)	Power at Cu (W)	Power at Ag (W)	Thickness of the film (μm)
TiCu	48:52	150	29	0	1.48
TiCu-10% Ag	43:47	143	27	3	1.44
TiCu-20% Ag	38:42	128	24	5	1.43
TiCu-30% Ag	33:37	113	21	9	1.46